ISOTHERMAL AND ISOCHRONAL ANNEALING OF GAMMA-RAY IRRADIATED n-CHANNEL POWER VDMOS TRANSISTORS

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Abstract. The behaviour of densities of the oxide trapped charge and the interface traps in gamma-ray irradiated n-channel power VDMOS transistors during isothermal and isochronal annealing has been investigated. The experimental results have revealed the existence of latent interface trap buildup (LITB) process. By using of numerical modeling, based on the H-W model, the LITB process during isothermal annealing has been successfully simulated. The interface trap densities have been determined by both the midgap and the charge pumping methods, and results have shown good qualitative agreement between these two methods.

Key words: interface traps, oxide trapped charge, irradiation, annealing, MOS transistors

1. INTRODUCTION

As the consequence of the influence of ionizing radiation on metal-oxide-semiconductor (MOS) transistors, the positive trapped charge in the gate oxide and the interface traps at Si/SiO₂ interface are formed [1]. To understand the behaviour of these defects is very important, since they can degrade the electrical performance of MOS transistors: threshold voltage, transconductance, leakage current and breakdown voltage. Also, the investigation of the radiation defects is very important for the ionizing radiation dosimetry, which can be based on p-channel MOS (pMOS) dosimetric transistors [2-7].

The particular attention should be given to the annealing of irradiated transistors, because of possible existence of latent interface trap buildup (LITB) process [8-10]. LITB process is defined as a sudden rapid increase in interface traps during postirradiation annealing after apparent saturation of the interface trap density immediately following

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irradiation [8,9]. Compared to the time scale of normal (conventional) interface trap buildup, i.e. interface trap buildup that occurs during and immediately after irradiation, the time scale of LITB process is much longer. The LITB process is not yet completely understood, although recent investigations [11-14] of hydrogen cracking in the gate oxide of irradiated MOS transistors have helped understanding this process.

Our recent investigations [15-19] have shown the existence of the LITB process in gamma-ray irradiated n-channel power vertical double-diffused metal-oxide-semiconductor (VDMOS) transistors during annealing at relatively lower temperatures (room-temperature, 55 and 140 °C). These experimental results have been explained by the hydrogen-water (H-W) model [15,16]. The aim of this paper is investigation of the behaviours of oxide trapped charge density and interface trap density during annealing at higher temperature values. Also, by performing the isochronal annealing experiment, the validity of the H-W model is further checked.

2. EXPERIMENTAL DETAILS

The devices used in these experiments were commercial n-channel power VDMOS transistors EFP8N15, manufactured by Ei-Microelectronics, Ni {the transistor consists of 5174 hexagonal cells). The maximum voltage of these devices is 150 V, and maximum current is 8 A. The effective channel length is 2.6 µm, and the channel width of a single cell is 62.4 µm (for more detailed description of the hexagonal cell see Refs. [19,20]). The starting material was 400 µm thick n-type Si wafers with (100) orientation, on which the n-epi layer of about 14 µm was grown. After implantation of the p+-body region, the gate oxide with nominal thickness of 100 nm was grown in dry oxygen at 1100 °C, and annealed in nitrogen at 1100 °C for 20 min. There followed the deposition of 800 nm thick poly-Si gate, implantation of the p-channel and n+-source regions, and deposition of 1 µm thick CVD oxide. After metalization and post-metalization anneal at 550 °C in nitrogen, devices were mounted in standard plastic TO-220 packages.

The transistors were irradiated at room temperature using a 60Co source at a dose rate of 0.01 Gy(Si)/s, to the total dose of 150 Gy (Si). The irradiation was performed in the Metrology Laboratory of the Institute for nuclear science 'Vin~a', Vin~a-Belgrade. During irradiation, the gate bias (V_G) was kept at 10 V. After irradiation, the transistors were annealed in air at different temperatures (140, 200, 250 and 290 °C) with V_G = 10 and -10 V, using the Heraeus HEP2 system of temperature chambers with high temperature stability. During both the irradiation and annealing, the source and drain terminals were grounded.

Transistor subthreshold and transfer characteristics in saturation were measured at room-temperature. Equipment used for electrical characterisation consisted of a HP 8116A function generator and Keithley 237 Source-Measure-Unit (SMU), both controlled by a PC (Fig. 1). Namely, using HP 8116A the DC voltage was applied to the transistor gate (G), and the drain (D) was biased (V_D = 10 V) and drain-source current (I_{DS}) measured by the Keithley 237 SMU (the source was grounded). The threshold voltage was determined as the intersection between V_G-axis and the extrapolated linear region of
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The (I_DS)_{1/2} - V_G curve. The midgap-subthreshold (MG) method [21] was used to determine the areal densities of positive gate oxide trapped charge (∆N_{ox} [cm^{-2}]) and interface traps (∆N_{it} [cm^{-2}]).

The charge-pumping (CP) method [22,23,20] for determination of ∆N_{it} was also used. Using HP 8116A function generator, the triangular pulses with the frequency of f = 100 kHz, duty cycle 50 % and amplitude of ∆V_A = 3 V were applied to the transistor gate, while the source was grounded. The CP current (I_{cp}) was measured at drain terminals by the Keithley 237 SMU. The measurement system also contains the PC with interface card under control of a Turbo Pascal program (Fig. 2).

![Fig. 1. Block diagram of the experimental set-up for MG measurements.](image)

![Fig. 2. Block diagram of the experimental set-up for CP measurements.](image)

3. THE HYDROGEN-WATER (H-W) MODEL

In our previous papers [15-19], it has been shown that the H-W model can describe the LITB process during annealing of irradiated MOS transistors. In the following text, the most important characteristics of the H-W model will be briefly described. The H-W model starts of the fact that a small concentration of H⁺ ions exists at the Si/SiO₂ interface after long annealing time. One of the possible sources of H⁺ ions are H₂ molecules, which arrive from the adjacent structures (CVD oxide and poly-Si gate) to the gate oxide. Namely, H₂ molecules are released in CVD oxide or/and poly-Si gate, and slowly diffuse towards the interface (see the structure of used transistors [19]). It has been confirmed that the diffusion constants of hydrogen are decreased in poly-Si gate [24] and irradiated thermal oxide [25]. In the vicinity of the interface, these molecules are cracked at positively charged centers CC⁺ in the oxide, forming hydrogen ions [11]:

\[
CC^+ + H_2 \xrightarrow{k} CC - H + H^+.
\]

The H⁺ ions, formed in reaction (1), drift towards the interface under positive electric field in the oxide.³

Another possibility is that H⁺ ions, formed during irradiation, are captured by oxygen vacancies and represent gate oxide trapped charge [10,26]. During annealing they are liberated and drift towards the interface. However, the contribution of captured hydrogen

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¹ The least square method performed by MATLAB program package was used.
² In following text "interface".
³ It is supposed that the positive bias is applied to the gate.
ions to $\Delta N_{ot}$ should be relatively small, since it has been shown that almost all of oxide trapped charge are E' centers [27,28]. Fleetwood et al. [10] have supposed that captured H$^+$ ions can make about 10-20 % of net oxide charge density. Also, H$_2$O molecules can be responsible for the beginning of the LITB process [18].

When H$^+$ ion arrives at the interface, it picks up an electron from the substrate [29]:

$$H^+ + e^- \quad \text{(from the substrate)} \quad \xrightarrow{k_2} \quad H^0.$$  \hspace{1cm} (2)

Liberated highly reactive H$^0$ atom may react without an energy barrier [30,31] in one of the following manners:

(i) The creation of the interface trap Si$^*_0$, when H$^0$ reacts with an interface trap precursor Si$_i$-H [29]:

$$Si_i - H + H^0 \xrightarrow{k_1} \quad Si^*_0 + H_2.$$  \hspace{1cm} (3)

The P$_{b0}$ and P$_{b1}$ centers, which exist at (100) interface, represent interface traps [32,33]. The P$_{b0}$ defect has the structure: $\equiv Si_{i0}$, while the structure of the P$_{b1}$ defect, here denoted as $\equiv Si_{i1}$, is not yet known (remaining three bonds of $\equiv Si_{i1}$ defect can be bonded to various species). It can be proposed that either the $\equiv Si_{i0}$-H defect, precursor of the P$_{b0}$ center, or some hypothetical $\equiv Si_{i1}$-H defect, precursor of the P$_{b1}$ center, are responsible for interface trap increase during annealing. In the following text, since remaining three bonds of P$_{b0}$ and P$_{b1}$ defects are not significant, these centers will be denoted as Si$^*_0$, and their precursors as Si$_{i0}$-H.

It should be noted that Kim and Lenahan [28] have shown that the P$_{b0}$ centers are largely responsible for the behaviour of interface traps and that the density of radiation-induced P$_{b1}$ can be neglected. Oppositely, the results of some investigations [34,31] have shown that other center or centers probably exist at the interface and act as interface traps. However, since structure and behaviour of these centers haven't been given, they will not be considered in this analysis.

(ii) The passivation of the interface trap, when H$^0$ reacts with previously formed (fabrication process-, radiation- or annealing-induced) interface trap [30,35]:

$$Si^*_0 + H^0 \xrightarrow{k_3} \quad Si_i - H.$$  \hspace{1cm} (4)

(iii) Dimerization of hydrogen, when H$^0$ reacts with another H$^0$ also existing near the interface [36,37]:

$$H^0 + H^0 \xrightarrow{k_4} \quad H_2.$$  \hspace{1cm} (5)

It should be noted that the reaction (3) is the most probable at the start of the LITB process [18].

The H$_2$ molecules formed in reaction (3) diffuse towards the bulk of the gate oxide and can be cracked at CC$^+$ centers (reaction (1)). So, the reaction (1) represents the

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4 The point denotes unpaired electron, i.e. "uncoupled spin".
5 $\equiv Si_{i0}$ defect denotes silicon atom at the oxide-silicon interface back bonded to three silicon atoms from substrate and is elsewhere represented as $Si_{i3} \equiv Si_{i0}^*$. 
continuous source of H⁺ ions during long time annealing. These ions drift towards the interface to form interface traps. It means that only a small initial concentration of H⁺ ions is needed for the start of LITB process. The remaining H⁺ ions necessary for formation of maximum ∆Nₙ (∆Nₙ,max) are produced during the LITB process itself in reaction (1) through the repetitive reaction sequence (1) ⇒ (2) ⇒ (3) ⇒ (1) or (1) ⇒ (5) ⇒ (1). Also, the reaction (1) is an additional mechanism for neutralisation of oxide trapped charge, as well as the explanation for the dependence of ∆Nₙ increase rate on temperature (see the following text).

Reactions (1)-(5) can explain the interface trap creation. However, for the explanation of experimentally observed decrease of ∆Nₙ at very late annealing times, it is necessary to include at least one additional reaction. According to the H-W model, interface trap passivation process can be explained by the following reaction of H₂O molecules at the interface [18]:

\[
\text{Si}^* + \text{H}_2\text{O} \xrightarrow{k_7} \text{Si} - \text{OH} + \text{H}^+. \tag{6}
\]

It would be expected [18] that the H₂O concentration will first increase, up to its maximum value, and than slowly decrease. One possible description of this would be: [H₂O] = k₇t exp(-a tⁿ), where k₇, a and n are constants, and t is annealing time.

4. EXPERIMENTAL RESULTS AND DISCUSSION

Figure 3 shows the changes in ∆Nₙ and ∆N₀ during annealing at various temperatures with V₉ = 10 V. As it can be seen, these changes very much depend on the annealing temperature. The main characteristics of these experimental results are: 1) very great values of ∆Nₙ,max created during the LITB process, 2) the explicit dependence of creation rate of interface traps on temperature - ∆Nₙ increase rate rises with temperature (note the log-scale for time axis), and 3) the decrease of ∆Nₙ after long annealing times. Also, the correlation between the increase of ∆Nₙ and the decrease of ∆N₂₀ is observed at all investigated temperatures.

In previous papers [17,18], the numerical simulation has shown that hydrogen ion transport H⁺ model [29,38,39] cannot explain this behaviour. It has been supposed [17] that the H⁺ model contains reactions (2)-(5), although only reactions (2) and (3) have originally been included in this model [29]. However, reactions (4) and (5) have been added to the H⁺ model since H⁺ atoms formed in (2) can react in these manners without an energy barrier [36,37,30,35], so that the probabilities for these reactions are great. It should be noted that neither the original H⁺ model nor the H⁺ model with reactions (4) and (5) added could fit the experimental results [17,18]. However, the H⁺ model (i.e. reactions (2) and (3)) is foundation of all models proposed for LITB process so far. The main difference between these models is the origin of H⁺ ions and the moment at which they reach the interface, but all models assume that interface traps are formed in reactions (2) and (3) [9,10,26]. In the case of normal interface trap buildup, for which the H⁺ model was originally proposed [29,38,39], the H⁺ ions arrive at the interface during time interval ranging from 1 to 10⁴ s [40,38] after pulsed irradiation, which is much shorter than time interval in which LITB occurs (even up to 10⁷ s [10,18]). According to recently proposed LITB process models [9,10], the H⁺ ions are created in the cracking reaction of H₂.
molecules in the oxide, after diffusion of H₂ from the structures adjacent to the gate oxide. The H⁺ than drifts to the interface to take part in reactions (2) and (3). Finally, the latest model [26] for LITB process assumes the capture of radiation-induced H⁺ ions at O vacancies and their release and drift towards the interface at later annealing times.

In the case of higher annealing temperatures (Fig. 3), the LITB process, which is in correlation with the latent ΔNₘ decrease, appears earlier, because of the faster diffusion of H₂ molecules from adjacent structures and earlier liberation of captured H⁺ ions. The creation rate of interface traps, i.e. the slope of ΔNᵣ curve during the LITB process, increases with temperature, as a consequence of the dependence of reaction (1) on temperature (activation energy of this reaction is approximately 1 eV [11,18]). Also, in the case of higher temperature, the time to beginning of passivation process is shorter, because of the faster diffusion of H₂O molecules to the interface.

![Fig. 3. ΔNᵣ and ΔNᵣ during annealing with V₆ = 10 V at different temperatures.](image1)

![Fig. 4. ΔNᵣ and ΔNᵣ during annealing with V₆ = -10 V at different temperatures.](image2)

It should be also noted that ΔNᵣ,max is lower in the case of a higher temperature. One of the reasons for such behaviour can be the existence of passivation reactions between H₂ molecules and interface traps above 200 °C [41]:

$$\text{Si}^* + \text{H}_2 \rightarrow \text{Si} - \text{H} + \text{H}^+.$$  

This reaction decreases the number of both the H₂ molecules, which diffuse towards bulk of the oxide, and interface traps. Other possibility is that the H₂O molecules, because of the faster diffusion, arrive at the interface earlier and contribute to the passivation process before significant density of interface traps is formed.
Figure 4 shows the changes in $\Delta N_{it}$ and $\Delta N_{it}$ during annealing at various temperatures with $V_G = -10$ V. As can be seen, $\Delta N_{it,max}$, as well as the time to beginning of the LITB process depend on annealing temperature as in the case of $V_G = 10$ V, but the values of $\Delta N_{it}$ are considerably lower (see Fig. 3). The LITB process may seem unexpected, since oxide electric field is negative and H$^+$ ions are prevented to arrive at the interface. However, the explanation for such behaviour lies in a two-stage nature of the reaction (1) [11]. Namely, in the first stage $H^0$ atom is created:

$$CC^+ + H_2 \xrightarrow{k_1} CC - H^+ + H^0.$$

and the H$^+$ ion is a product of the second stage of this reaction (see Ref. [11]). Formed $H^0$ atom can leave the place of creation and diffuse towards the interface to create the interface trap in reaction (3). The reasons for dependence of both the creation rate of $\Delta N_{it}$ and values of $\Delta N_{it,max}$ on temperature are the same as at $V_G = 10$ V.

### 4.1 Numerical simulation

Besides the above qualitative analysis of the behaviour of interface traps, the numerical modeling of their kinetics can be performed. Namely, on the basis of bimolecular theory [42] and reactions (1)-(6), the following differential equations are obtained for the H-W model:

$$\frac{d[H^+]}{dt} = k_1[(CC^+)[H_2] - k_2[H^+][e^-]],$$

$$\frac{d[Si^*]}{dt} = k_3[Si, H[H^0] - k_4[Si^*][H^0] - k_6[Si^*][H_2O]),$$

$$\frac{d[H^0]}{dt} = k_2[H^+][e^-] - (k_3/t_{ox})[Si, H][H^0] - (k_4/t_{ox})[Si^*][H^0] - k_6[H^0]^2 + (k_6/t_{ox})[Si^*][H_2O]$$

$$\frac{d[H_2]}{dt} = -k_1[(CC^+)[H_2] + (k_3/t_{ox})[Si, H][H^0] + k_3[H^0]^2,$$

$$\frac{d[Si, H]}{dt} = -k_3[Si, H][H^0] + k_4[Si^*][H^0],$$

$$\frac{d(CC^+)}{dt} = -k_1[(CC^+)[H_2],$$

$$\frac{d[e^-]}{dt} = 0.$$

Reactions (9)-(15) describe the kinetics of interface traps in irradiated transistors during annealing with positive gate bias. The numerical fourth-order Runge-Kutta method with adaptive step-size [43] was used to solve this system of coupled differential equations. The set of unknown reaction rate constants and starting values of unknown reagent concentrations was found as described in [17].

Figure 5 gives the results of numerical simulation for annealing with $V_G = 10$ V at
various temperatures. The parameters of simulation at 140 °C are: \([\text{Si}^+_0]_0 = \Delta N_{\text{it}}(0) = 4.5 \times 10^{10} \text{ cm}^{-2}\) (\(\Delta N_{\text{it}}(0)\) is the interface trap density at the start of annealing), \([\text{H}^+]_0 = 10^{14}\text{ cm}^{-3}\), \([\text{H}_2]_0 = 0\), \([\text{CC}]_0 = 3.25 \times 10^{14}\text{ cm}^{-3}\), \([\text{SiH}]_0 = 5 \times 10^{12}\text{ cm}^{-2}\), \([\text{e}^-]_0 = 3.5 \times 10^{13}\text{ cm}^{-3}\), \(k_1 = 10^{15}\text{ cm}^{3}\text{h}^{-1}\), \(k_2 = 2.4 \times 10^{15}\text{ cm}^{3}\text{h}^{-1}\), \(k_3 = 10^{15}\text{ cm}^{3}\text{h}^{-1}\), \(k_4 = 0.4 \times k_3\), \(k_5 = 0.09 k_3\), \(k_6 = 2 \times 10^{14}\text{ cm}^{3}\text{h}^{-1}\), \(k_7 = 10 \text{ cm}^{3}\text{h}^{-1}\), \(n = 0.08\), \(a = 8.5\). H\(_2\) molecules, which are cracked at CC\(^+\) centers, and H\(_2\)O molecules need some time to arrive to the interface: \(t_{\text{lat}}\) and \(t_{\text{lat}1}\), respectively [18]. For this temperature: \(t_{\text{lat}} = 0.15\text{ h}\), and \(t_{\text{lat}1} = 2\text{ h}\), while the initial concentration of delayed H\(^+\) ions which arrive at the interface is \(c = 2.9 \times 10^{16}\text{ cm}^{-3}\).

Different parameters at 200 °C, than at 140 °C, are: \([\text{CC}]_0 = 2.5 \times 10^{16}\text{ cm}^{-3}\), \(k_1 = 6.3 \times 10^{14}\text{ cm}^{3}\text{h}^{-1}\), \(k_2 = 2 \times 10^{13}\text{ cm}^{3}\text{h}^{-1}\), \(k_6 = 4.2 \times 10^{14}\text{ cm}^{3}\text{h}^{-1}\), \(n = 0.138\), \(t_{\text{lat}} = 0.12\text{ h}\), \(t_{\text{lat}1} = 0.126\text{ h}\), \(c = 1.8 \times 10^{16}\text{ cm}^{-3}\), for 250 °C: \([\text{CC}]_0 = 2.1 \times 10^{16}\text{ cm}^{-3}\), \(k_1 = 1 \times 10^{12}\text{ cm}^{3}\text{h}^{-1}\), \(k_2 = 8 \times 10^{15}\text{ cm}^{3}\text{h}^{-1}\), \(k_6 = 1.5 \times 10^{12}\text{ cm}^{3}\text{h}^{-1}\), \(n = 0.21\), \(a = 9.65\), \(t_{\text{lat}} = 0.045\text{ h}\), \(t_{\text{lat}1} = 0.0045\text{ h}\), \(c = 0.85 \times 10^{16}\text{ cm}^{-3}\) and for 290 °C: \([\text{CC}]_0 = 1.87 \times 10^{16}\text{ cm}^{-3}\), \(k_1 = 6.1 \times 10^{12}\text{ cm}^{3}\text{h}^{-1}\), \(k_2 = 2 \times 10^{12}\text{ cm}^{3}\text{h}^{-1}\), \(k_6 = 1.5 \times 10^{11}\text{ cm}^{3}\text{h}^{-1}\), \(n = 0.074\), \(a = 8.32\), \(t_{\text{lat}} = 0.034\text{ h}\), \(t_{\text{lat}1} = 0.003\text{ h}\), \(c = 0.95 \times 10^{16}\text{ cm}^{-3}\).

On the basis of above presented results of numerical simulation, the activation energy for some reactions can be determined. Namely, using the Arrhenius equation:

\[
k = k_0 \exp \left(-\frac{E_a}{k_B T} \right)
\]

(16)

where \(k_B\) is the Boltzmann's constant, the values of the activation energies are obtained for reaction (1): \(E_a = 1.17\text{ eV}\) (this value is in good agreement with \(E_a = 1\text{ eV}\) [11] and \(E_a = 1.1\text{ eV}\) [18]); reaction (2): \(E_a = 0.9\text{ eV}\) and reaction (6): \(E_a = 1.5\text{ eV}\). Figure 6 shows the dependencies of natural logarithms of reaction rate constants \(k_1\), \(k_2\) and \(k_6\) on reciprocal temperature.

Fig. 5. \(\Delta N_{\text{it}}\) during annealing with \(V_0 = 10\text{ V}\) at different temperatures. The curves represent the results of numerical modeling based on the H-W model.

![Fig. 5](image-url)

Fig. 6. Dependencies of reaction rate constants \(k_1\), \(k_2\) and \(k_6\) on reciprocal annealing temperature.

![Fig. 6](image-url)

\(^6\) The relations between reaction rate constants \(k_3\), \(k_4\) and \(k_5\) are obtained on the basis of the radius captures of the reactants of corresponding reactions [44,18].
Figure 7 gives the results of numerical simulation for annealing with $V_G = -10$ V at various temperatures. The differential equations for modeling of annealing under negative bias are similar to the system (9)-(15). However, since the reaction (2) doesn’t exist in the case of negative bias, it means that there are no equations (9) and (15). Also, $k_1$ should be substituted with $k'_1$, and the term $k_2[H^+][e^-]$ in the equation (11) with the term $k'_1[CC^+][H_2]$.

Different parameters for annealing with $V_G = -10$ V than with $V_G = 10$ V, for $t = 140$ °C, are: $[CC^+]_0 = 1.85 \times 10^{16}$ cm$^{-3}$ for $t = 200$ °C, $[CC^+]_0 = 1.4 \times 10^{16}$ cm$^{-3}$, $k'_1 = 8 \times 10^{-15}$ cm$^3$/h, and for $t = 250$ °C: $[CC^+]_0 = 1.2 \times 10^{16}$ cm$^{-3}$, $k'_1 = 9 \times 10^{16}$ cm$^3$/h, $k_6 = 1.6 \times 10^{-12}$ cm$^3$/h, $n = 0.175$, $a = 10$.

4.2 Charge pumping (CP) method

To check the above shown experimental results for $\Delta N_{it}$, obtained by the MG method, the CP method was also used. Figure 8 shows $\Delta N_{it}$ during annealing at various temperatures with $V_G = 10$ V, obtained by CP method. The average interface trap densities are determined as $\Delta N_{it} = N_{it}(t) - N_{it0}$, where $N_{it}(t)$ is the absolute interface trap density after annealing time $t$, and $N_{it0}$ is the absolute interface trap density before irradiation. These interface trap densities are determined in the following manner [22,23]:

$$\Delta N_{it} = \frac{D_{it}}{\frac{I_{cp,max}}{q A_G}} \times \Delta E$$

where $D_{it}$[cm$^{-2}$eV$^{-1}$] is the average energetic interface trap density, $\Delta E$ is the energetic range, $I_{cp,max}$ is the maximum value of the CP current, $q$ is the absolute value of the electron charge, and $A_G = 6.163 \times 10^{-6}$ cm$^2$ × cell number [20] is area under the gate active in charge-pumping.

Figure 9 gives $\Delta N_{it}$ during annealing at 200 °C and $V_G = \pm 10$ V, obtained by MG and CP methods. On the basis of this figure, and the comparison between Figs. 3 and 8, it can be concluded that the qualitative behaviours of $\Delta N_{it}$ obtained by both methods are the same, but the values of $\Delta N_{it}$ are significantly lower in the case of CP method. This qualitative agreement between the experimental results obtained by MG and CP methods confirms our previous results [17,18], i.e. the existence of the LITB process and its dependence on temperature. The sources of quantitative differences in $\Delta N_{it}$ values may be different widths and/or positions of energetic range $\Delta E$ within silicon band gap, in which the interface traps are registered by CP and MG methods. In the case of CP method, we have determined the energetic range of $\Delta E = 0.466$ eV [45], centered around midgap.
In the case of MG method, the average density of interface traps is determined in the range from midgap, where surface potential is \( \psi_s = \psi_B = 0.381 \text{ V} \), to the value of surface potential of \( \psi_B = \psi_T \), which corresponds to the threshold voltage \( V_T \). Since for \( V_G = V_T \) the surface potential has values between \( \psi_{T1} = 2\psi_B \) and \( \psi_{T2} = 2\psi_B + 6U_T \) \( (U_T = k_BT/q = 0.02587 \text{ V}) \) [46], it can be concluded that energetic range \( \Delta E \) has value between \( \Delta E_1 = q(\psi_{T1} - \psi_B) \) and \( \Delta E_2 = q(\psi_{T2} - \psi_B) \), i.e. 0.381-0.536 eV. For MG method, we have estimated the value of \( \Delta E = 0.424 \text{ eV} \) [45], which is very close to the value obtained for CP method (0.466 eV). It means that the disagreement in values of \( \Delta N_{it} \) obtained by these methods cannot be the consequence of difference in \( \Delta E \), but rather the consequence of different portions of band gap in which \( \Delta N_{it} \) values are determined (i.e. positions of \( \Delta E \) within band gap). Namely, since the distribution of interface traps within band gap of silicon is not uniform, but has the 'U'-shape [1], these portions within band gap are very important. The density of interface traps is greater at the band gap edges (accessible to MG method), and is lower around midgap (accessible to CP method).

It should be also noted that CP method gives \( \Delta N_{it} \) predominantly for the SiO2/n-epi interface, while MG method senses interface traps at the channel interface (see the cross section of used transistors [19]), what can also contribute to observed quantitative disagreements [47]. Finally, owing to a great difference in the effective frequencies of two techniques, different amounts of 'border' traps contribute to signals measured by them, leading to an overestimation of the number of 'true' interface traps by the midgap method [48].

### 4.3 Activation energy of the LITB process

Activation energy \( E_a \) of the LITB process can be determined by Arrhenius equation:

\[
E_a = k_B T \ln \left( \frac{N_A}{n_i} \right)
\]

where \( N_A \) is the concentration of acceptor ingredient in semiconductor and \( n_i \) is the intrinsic concentration.
where $t$ is the annealing time, and $t_o$ is normalizing constant. Figure 10 shows the dependence of time for half of latent $\Delta N_{it}$ increase $(t_{1/2})$ on reciprocal temperature $(1/T)$ for four annealing temperatures (140, 200, 250 and 290°C) for EFP8N15 type transistors and for two temperatures (room-temperature and 55°C) for the similar EFL1N10 type transistors [18] ($V_G = 10$ V for all transistors during annealing). The fitting through all experimental values gives the activation energy of $E_{a1} = 0.63$ eV, and division to two temperature ranges: $E_{a2} = 0.54$ eV (room-temperature - 140°C) and $E_{a3} = 0.47$ eV (200 - 290°C). In this figure, it can be seen that more precise fitting is performed in the case of two temperature ranges. The obtained values of $E_a$ are probably in relation with activation energy of diffusion of H$_2$ molecules ($E_a \approx 0.45$ eV). This is in agreement with the H-W model in which H$_2$ molecules diffusing from structures adjacent to gate oxide towards the interface, are responsible for the LITB process.

4.4 Isochronal annealing experiments

$\Delta N_{it}$ and $\Delta N_{it}$ obtained by MG and CP methods during isochronal annealing are presented in Fig. 11. The transistors were kept at each temperature for $\tau = 5$ min, the temperature interval between two annealing points was $\Delta t = 10$ °C, and gate bias was $V_G = 10$ V. As can be seen, the values of $\Delta N_{it}$ and $\Delta N_{it}$ are almost constant up to temperature of 175°C. After this temperature, the sudden increase of $\Delta N_{it}$, accompanied by the sudden decrease of $\Delta N_{it}$, is observed. It is one more evidence that formation of interface traps is in correlation with the annealing of the oxide trapped charge. $\Delta N_{it}$ starts to decrease above 225°C. As can also be seen, the qualitative behaviours of $\Delta N_{it}$ obtained by MG and CP methods are the same, but the values of $\Delta N_{it}$ are significantly lower in the case of CP method.

On the basis of data in Fig. 11, using the model proposed by Danchenko et al. [49], which has been widely used in the literature [50-52], the activation energy distribution of the annealing of oxide trapped charge was determined. Namely, using appropriate approximations, it can be shown [49] that the activation energy $\varepsilon$ is given by the transcendental equation:

$$
\frac{\varepsilon}{\eta} + \ln\left(\frac{\varepsilon}{\eta} + 2\right) = \ln(\lambda \eta),
$$

where $\eta = \kappa T$, $A$ is the constant called 'frequency factor', and $c = \tau/(\kappa T \Delta T)$ is the tempering constant. The initial distribution, i.e. distribution of oxide trapped charge
before annealing, in activation energies \(n_0(\varepsilon)\) is [49]:

\[
n_0(\varepsilon) = \frac{dN}{d\varepsilon} \frac{1}{\varepsilon/\eta + 1},
\]

(20)

where \(N\) is unannealed fraction of oxide trapped charge, defined as [49,50]:

\[
N(t) = \frac{\Delta N_{ot}(t)}{\Delta N_{ot}(0)},
\]

(21)

where \(\Delta N_{ot}(0)\) is the oxide trapped charge density after irradiation (before annealing), and \(\Delta N_{ot}(t)\) is this density at the moment \(t\) during annealing.

The procedure of finding of initial energy distribution \(n_0(\varepsilon)\) is the following. The values of activation energy for each temperature of isochronal annealing are found first by the transcendental equation (19) (in this paper, the iteration method was used to solve this equation [43]). After that, the values of \(dN/d\eta = (1/k_B)dN/dT\) derivative are numerically determined [43] on the basis of the dependence of unannealed fraction \(N\) on \(\eta\), i.e. on \(T\) (\(N\) has the same shape as \(\Delta N_{ot}\), shown in Fig. 11).

Figure 12 presents the initial distribution of oxide trapped charge in activation energies \(n_0(\varepsilon)\) for three values of the frequency factor \(A\). It can be concluded that the majority of oxide trapped charge has the same value of activation energy (sharp peaks in Fig. 12). However, distribution \(n_0(\varepsilon)\) is very sensitive to the value of constant \(A\), so that one should be very cautious when discussing \(E_a\) values obtained by Danchenko's method. The obtained energy distributions are even qualitatively different from those found from TSC measurements by Fleetwood et al. [53] in a wide range of oxides.

5. CONCLUSIONS

The LITB process has been observed in the wide range of temperatures during annealing of gamma-ray irradiated n-channel power VDMOS transistors. The experimental results are in consistence with the H-W model. The numerical simulation of
the LITB process, based on the H-W model, has been successfully performed. The MG and CP methods have been used for determination of interface trap densities, and have yielded qualitatively similar results. Observed quantitative differences have been discussed. On the basis of isothermal annealing, the activation energy of the LITB process has been determined, showing that H₂ molecules are probably responsible for this process. The experimental results of isochronal annealing have confirmed that the increase in ∆Nᵢ is in correlation with the decrease in ∆Nₓ. Application of Danchenko's model combined with electrical measurement technique for determination of oxide trap charge energy distribution has been critically discussed.

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Isothermal and Isochronal Annealing of Gamma-Ray Irradiated n-Channel Power VDMOS Transistors


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IZOTERMALNO I IZOHRONALNO ODŽARIVANJE
n-KANALNIH VDMOS TRANZISTORA SNAGE
OZRAČENIH GAMA-ZRAČENJEM

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Istpitivano je ponašanje gustina zahvaćenog nanelektrisanja i površinskih stanja kod ozračenih n-kanalnih VDMOS tranzistora snage tokom njihovog izotermalnog i izohronalnog ožarivanja. Eksperimentalni rezultati su potvrdili postojanje procesa latentnog formiranja površinskih stanja (LITB proces). Izvršeno je uspešno numeričko modelovanje ovog procesa zasnovano na H-W modelu. Gustine površinskih stanja su određivane pomoću midgap i charge-pumping metode, i pokazano je dobro kvalitativno slaganje vrednosti dobijenih ovim metodama.

Ključne reči: površinska stanja, zahvaćeno nanelektrisanje u oksidu, ožarivanje, ožarivanje, MOS tranzistor